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	Application No.	Applicant(s)	Al
Madian of Allamatica.	09/868,620	MATSUI ET AL.	
Notice of Allowability	Examin r	Art Unit	
	Peter J Lish	1754	
Th MAILING DATE of this communication apper All claims being allowable, PROSECUTION ON THE MERITS IS herewith (or previously mailed), a Notice of Allowance (PTOL-85) NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATENT RI of the Office or upon petition by the applicant. See 37 CFR 1.313	(OR REMAINS) CLOSED in t or other appropriate commun GHTS. This application is sul	his application. If not included ication will be mailed in due c	d :ourse. THIS
1. A This communication is responsive to <u>Preliminary Amendment</u>	ent, filed 3/30/04.		
2. The allowed claim(s) is/are 1-3 and 5-21.			
3. \boxtimes The drawings filed on <u>25 June 2001</u> are accepted by the Ex	xaminer.		
 4. ☐ Acknowledgment is made of a claim for foreign priority una) ☐ All b) ☐ Some* c) ☐ None of the: Certified copies of the priority documents have Certified copies of the priority documents have Copies of the certified copies of the priority documents have Copies of the certified copies of the priority documents have international Bureau (PCT Rule 17.2(a)). * Certified copies not received:	been received. been received in Application	No	on from the
Applicant has THREE MONTHS FROM THE "MAILING DATE" noted below. Failure to timely comply will result in ABANDONM THIS THREE-MONTH PERIOD IS NOT EXTENDABLE.		reply complying with the requ	uirements
5. A SUBSTITUTE OATH OR DECLARATION must be submi			OTICE OF
6. CORRECTED DRAWINGS (as "replacement sheets") mus	t be submitted.		
(a) ☐ including changes required by the Notice of Draftspers	on's Patent Drawing Review (PTO-948) attached	
1) ☐ hereto or 2) ☐ to Paper No./Mail Date			
(b) ☐ including changes required by the attached Examiner's Paper No./Mail Date	s Amendment / Comment or in	the Office action of	
Identifying indicia such as the application number (see 37 CFR 1. each sheet. Replacement sheet(s) should be labeled as such in the			back) of
7. DEPOSIT OF and/or INFORMATION about the deposit attached Examiner's comment regarding REQUIREMENT			ote the
Attachment(s)			
1. Notice of References Cited (PTO-892)	5. Notice of Info	rmal Patent Application (PTO	-152)
2. Notice of Draftperson's Patent Drawing Review (PTO-948)	6. ☐ Interview Sun Paper No./M	nmary (PTO-413), ail Date	
 Information Disclosure Statements (PTO-1449 or PTO/SB/0 Paper No./Mail Date <u>1/7/04</u> 	8), 7. 🗌 Examiner's A	mendment/Comment	
4. Examiner's Comment Regarding Requirement for Deposit	<u></u>	atement of Reasons for Allov	vance
of Biological Material	9.	STANLEY S SILVERM SUPERVISORY PATENT EX	

U.S. Patent and Trademark Office PTOL-37 (Rev. 1-04) TECHNOLOGY CENTER 1700

Preliminary Amendment U.S. Application S. N. 09/868,620

Attorney Docket No. 010756

IN THE CLAIMS

Please cancel claim 4 without prejudice or disclaimer, and amend claims 1, 13 and 14 as

follows:

Claim 1 (Currently amended): Amorphous nano-scale carbon tubes each containing a main

framework which comprises carbon, and each having a straight shape, a diameter of 0.1 to 1000 nm

and an amorphous structure, and each having an interlayer spacing (002) between hexagonal carbon

layers of at least 3.7 Å, a diffraction angle (20) of 24.1 degrees or less, and a 20 band half-width of

at least 3.2 degrees, as determined with a diffractometer by an X-ray diffraction method (incident

X-Ray: CuKα).

Claim 2 (Previously Presented): Amorphous nano-scale carbon tubes according to claim 1,

each of which comprises hexagonal carbon layers each having a dimension of the planar direction

that is smaller than the diameter of the carbon tube, as determined from a transmission electron

microscope image.

Claim 3 (Previously Presented): Amorphous nano-scale carbon tubes according to claim 1,

each of which has a 20 band half-width of at least 7.0 degrees, as determined with a diffractomer by

an X-ray diffraction method (incident X-ray: CuKα).

Claim 4 (Canceled):

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Claim 5 (Previously Presented): Amorphous nano-scale carbon tubes according to claim 1, each of which has a hollow cylindrical shape or a hollow rectangular prism shape.

Claim6 (Previously Presented): Amorphous nano-scale carbon tubes according to claim 1, each of which has at least one open end.

Claim 7 (Previously Presented): Amorphous nano-scale carbon tubes according to claim 1, which are formed on a substrate, a particle or a porous material.

Claim 8 (Previously Presented): A gas-storing material comprising an amorphous carbonaceous material containing the amorphous nano-scale carbon tubes according to claim 1.

Claim 9 (Original): The gas-storing material according to claim 8, which contains at least one of a metal salt and a metal.

Claim 10 (Original) The gas-storing material according to claim 9, wherein the metal salt and the metal are selected from the group consisting of iron, cobalt, nickel, copper, platinum, palladium, rubidium, strontium, cesium, vanadium, manganese, aluminum, silver, lithium, potassium, sodium, magnesium, hydrogen-occluding alloys and metal complexes.

Claim 11 (Original) A method for storing a gas, wherein a gas is stored using the gas-storing material according to any one of claims 8 to 10.

Claim 12 (Original) The method according to claim 11, wherein the gas to be stored is hydrogen, methane, helium, neon, xenon, krypton or carbon dioxide.

Claim 13 (Currently amended): A method for producing a carbon material containing amorphous nano-scale carbon tubes according to claim 1, the method comprising subjecting a heat decomposable resin having a decomposition temperature of 200 to 900 °C to excitation treatment in the presence of a catalyst comprising a metal powder and/or a metal salt,

the heat decomposable resin being selected from the group consisting of polytetrafluoroethylene, polyvinylidene chloride, polyvinylidene fluoride and polyvinyl alcohol, and the catalyst being at least one halide of a metal selected from the group consisting of magnesium, iron, cobalt and nickel.

Claim 14 (Currently amended): The method for producing said carbon material containing the amorphous nano-scale tubes according to claim 13, wherein the catalyst is iron chloride metal powder and/or the metal salt is at least one member selected from the group consisting of alkaline earth metals, iron, cobalt, nickel, chromium and their salts.

Claim 15 (Original) The method for producing said carbon material containing the

amorphous nano-scale carbon tubes according to claim 13 or 14, wherein the excitation treatment

of the heat decomposable resin is carried out by a heat treatment in an inert gas at a temperature of

300 to 3000°C.

Claim 16 (Original) The method for producing said carbon material containing the

amorphous nano-scale carbon tubes according to claim 13 or 14, wherein the excitation treatment

of the heat decomposable resin is carried out by a light irradiation treatment in an inert gas at a

temperature of room temperature to 3000°C.

Claim 17 (Original) The method for producing said carbon material containing the

amorphous nano-scale carbon tubes according to claim 13 or 14, wherein the excitation treatment

of the heat decomposable resin is carried out by plasma treatment in an inert gas at a temperature of

room temperature to 3000°C.

Claim 18 (Original) The method for producing said carbon material containing the

amorphous nano-scale carbon tubes according to claim 13 or 14, wherein the excitation treatment

of the heat decomposable resin is carried out by electron beam irradiation treatment in an inert gas

at a temperature of room temperature to 3000°C.

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Claim 19 (Original) The method for producing said carbon material containing the amorphous nano-scale carbon tubes according to claim 13 or 14, wherein the excitation treatment of the heat decomposable resin is carried out by ion beam irradiation treatment in an inert gas at a temperature of room temperature to 3000°C.

Claim 20 (Previously presented): A carbon material containing the amorphous nano-scale carbon tubes according to claim 1.

Claim 21 (Previously presented): The amorphous nano-scale carbon tubes according to claim 1, each of which has an interlayer spacing (002) between hexagonal carbon layers of 3.9 to 4.7 Å, a diffraction angle (20) of 18.9 to 22.6 degrees, and a 20 band half-width of 7.6 to 8.2 degrees, as determined with a diffractometer by an X-ray diffraction method (incident X-ray: $CuK\alpha$).